CO-OCCURRENCE OF EREMOPHILANES AND EUDESMANES IN ARTEMISIA PECTINATA

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Abstract—The aerial parts of Artemisia pectinata afforded, in addition to a typical spiroketal enol ether and the known eremophil-9,11(13)-dien-12-oic acid, the corresponding 8,12-lactone as well as two eudesmane derivatives, a hydroxy and an acetoxy acid. The co-occurrence of eremophilanes and eudesmanes and the absence of the corresponding eudesman-8,12-olide with a trans-fused lactone ring is remarkable.

INTRODUCTION

The chemistry of the large genus Artemisia is not very uniform. In addition to different types of acetylenic compounds [1], coumarins [2], lignans [3] and sesquiterpene lactones [4] are widespread. Although eudesmanolides are very common in this genus, other types of sesquiterpene lactones have also been reported. We have now studied a species from Mongolia, Artemisia pectinata. Pall A species collected 300 km SW of Ulan-Bator afforded some sesquiterpenes, including eremophila-9(10),11(13)-dien-12-oic acid [5]. Our results on material collected ca 600 km NW of Ulan-Bator are discussed in this paper.

RESULTS AND DISCUSSION

The aerial parts of A pectinata afforded the typical spiroketal 1, which is present in only some groups of the Anthemideae [1], and as the main constituent the eremophilane derivative 2, which so far has been isolated only from Athanasia [6] and Baeria species [7] Furthermore, the corresponding lactone 3 and the eudesmane derivatives 4 and 5 were present

The structure of 3, molecular formula $C_{19}H_{20}O_2$, followed from the ¹H NMR spectral data (Table 1). Spin decoupling allowed the assignment of all signals Inspection of a model further showed that the couplings observed required the proposed relative stereochemistry

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Table 1	¹ H NMR	spectral data	of 3-6	(400 MHz,	CDCl ₃ ,	TMS	as
		interna	l standa	rd)			

	Me ester						
	3*	4	of 4	5	6		
H-6	1 82 dd	2 59 dd	2 58 dd	2 73 dd	3 01 dd		
H-6'	1 <i>77 dd</i>	1 99 dd	1 98 dd	1 89 dd	1 89 dd		
H-7	3 18 <i>ddddd</i>	2 40 ddd	2 38 ddd	2 65 ddd	2 29 ddddd		
H-8	4 87 br dd	3 94 ddd	3 96 ddd	5 30 ddd	4 12 ddd		
H-9	\	1 88 m	1 90 m	2 03 m	2 10 m		
H-9'	5 46 dd	1 55 m	1 55 m	1 57 m	1 50 dd		
H-13	6 26 4	6 42 s	6 29 s	6 36 s	6 09 d		
H-13'	5 60 d	5 77 s	5 70 s	573s	5 44 d		
H-14	095s	1 02 s	1 09 s	1 15 s	1 09 s		
H-15	0 87 d	1 58 br s	1 59 br s	1 62 br s	1 62 br s		
OAc				196s			
OMe	_		3 77 <i>s</i>	-			

*H-1 2 24 ddddd, H-1' 2 04 br d

J (Hz) Compound 3 1, 1' = 1, 2' = 13, 1, 2 = 5, 1, 8 = 1, 9 = 1, 4, 15 = 7, 6, 6' = 14, 6, 7 = 7, 6', 7 = 6, 7, 8 = 7, 7, 13 = 3, 7, 13' = 27,

at C-7 and C-8 A corresponding lactone, epimeric at C-7 and C-8, has previously been reported from a Frullania species [8] Furthermore, a 6-hydroxy derivative of 3 has been prepared from alantolactone [9] The ¹H NMR spectrum is very similar to that of 3

The ¹H NMR spectrum (Table 1) of the methyl ester of 4, obtained by addition of diazomethane, displayed signals of an \alpha-substituted acrylate which typically showed no allylic coupling A broadened singlet at $\delta 1$ 59 and the absence of further olefinic signals required a tetrasubstituted double bond, while a second methyl signal at δ 1 09 indicated the presence of a eudesmanolide Spin decoupling allowed the assignment of most signals and clearly showed that a hydroxyl group was at C-8 The configuration at this carbon followed from the couplings with H-7 and H-9 The structure was supported further by the ¹³C NMR spectrum (see Experimental) Finally 4 was transformed to the lactone 6 via the tosylate Again the ¹H NMR spectral data clearly supported the structure and the stereochemistry (Table 1) The latter was established further by the positive Cotton effect, which, following the Geissman rule [10], is an indication that an 8,12-trans-eudesmanolide was present Thus 6, which was also isolated as a natural compound, is the 8-epimer of a desoxy derivative of ivangustin [11] In addition to 6, a 5hydroperoxide of $6 (\Delta^{4,15})$ was obtained, which, however, most likely was an artefact as the vinyl proton signals were not present in the starting material. The structure followed from the ¹H NMR spectrum

The structure of 5 could be deduced from the spectroscopic data. While most data were similar to those of the methyl ester of 4, the ¹H NMR signal of H-8 was shifted downfield and a singlet at $\delta 1$ 96 indicated the presence of an acetate group

The chemistry of this Artemisia species is unusual and may support its placement in the new genus Neopallasia [12] Further investigation of more species may show whether these constituents are more widespread in this genus

EXPERIMENTAL

The air-dried aerial parts (600 g, collected in Mongolia in summer 1983, voucher deposited at the Academy Institute at Halle, GDR) were extracted with MeOH-Et₂O-petrol (1 1 1) and the extract obtained (22 g) was worked up in the usual way [13] CC (SiO₂) fractions were as follows 1 (Et₂O-petrol, 1.10), 2 (Et₂O-petrol, 1 4), 3 (Et₂O-petrol, 1 3) and 4 (Et₂O-petrol, 1 1, and Et₂O) Repeated CC (Et₂O-petrol, 1 10) of fraction 1 gave 4 mg 1 and 65 g 2 (compared with authentic material, 400 MHz 1H NMR and TLC) TLC (SiO2, PF 254, Et2O-petrol, 1 4) of fraction 2 afforded crude 3, which was purified by distillation (bp_{0.1Torr} $\sim 150^{\circ}$, bath temp, short distil) yielding 300 mg 3 Fraction 3 contained crude 5 TLC (Et₂O-C₆H₆, 1 1) of one tenth of this fraction gave 240 mg $5 (R_f 0.45)$ One tenth of fraction 4 was purified by TLC (Et₂O, R_f 0 42) affording 440 mg 4 A second investigation afforded, in addition to these compounds, 10 mg 6, identical with the compound obtained from 4

Eremophil-9,11(13)-dien-8β,12-olide (3) Colourless oil, bp_{0.1Torr} ~ 150°, IR $\nu_{\text{max}}^{\text{CCl}_*}$ cm⁻¹ 1770 (y-lactone), MS m/z (rel int) 232 146 [M]⁺ (58) (calc for C₁₅H₂₀O₂ 232 146), 217 [M-Me]⁺ (9), 204 [M-CO]⁺ (6), 136 (59), 121 [136-Me]⁺ (76), 91 (100)

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+47.9 \quad +54.3 \quad +59.4 \quad +73.6} (\mathrm{CHCl_3}, \ c \ 0.3)$$

8α-Hydrox yeudesma-4,11(13)-dien-12-oic acid (4) Viscous, colourless oil, 13 C NMR (CDCl₃) (C-1-C-15) 39 7, 18 7, 32 8*, 126 1, 142 0, 30 5*, 48 3, 70 6, 50 3, 35 8, 132 4, 171 9, 126 5, 25 4, 19 4 (Signals labelled with an * may be interchangeable) To 40 mg 4, CH₂N₂ in Et₂O was added After evapn, TLC (Et₂O-petrol, 1 1, 3 developments, R_f 0 34) gave 40 mg Me ester of 4, colourless oil, IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹ 3620 (OH), 1730, 1635 (C=CCO₂R), MS m/z (rel int) 264 173 [M]⁺ (11) (calc for C₁₆H₂₄O₃ 264 173), 246 [M-H₂O]⁺ (58), 232 [M-MeOH]⁺ (24), 231 [246-Me]⁺ (100), 217 [232-Me]⁺ (41), 199 [217-H₂O]⁺ (20), 171 [199-CO]⁺ (58), 91 (56)

To 50 mg 4 in 01 ml pyridine, 50 mg p-tosyl chloride was added After standing at 20° for 24 hr, usual work-up afforded by

^{8, 9 = 4}, compounds 4 and 5 6, 6' = 14, 6, 7 = 35, 6', 7 = 12, 7, 8 = 12,

^{8, 9 = 11, 8, 9&#}x27; = 4, compound 6 7, 13 = 31, 7, 13' = 28

TLC (Et₂O-petrol, 1 3, 3 developments, R_f 0 30) 30 mg 6, colourless oil, IR $v_{\text{max}}^{\text{CCl}}$ cm⁻¹ 1780 (γ -lactone), MS m/z (rel int) 232 146 [M]⁺ (24) (calc for C₁₅H₂₀O₂ 232 146), 217 [M - Me]⁺ (100), 123 (51), 91 (44), CD (MeCN) $\Delta \epsilon_{304} = -0.05$, $\Delta \epsilon_{261} = +0.18$

A second fraction (10 mg, R_f 0 20) contained the corresponding hydroperoxide (5-OOH of 6) (Δ^{4-15}), colourless oil, IR $v_{\max}^{\rm CCL}$ cm⁻¹ 3520 (OOH), 1780 (γ -lactone), MS m/z (rel int) 231 [M – OOH]⁺ (29), 149 (53), 95 (77), 53 (100)

 8α -Acetoxyeudesma-4,11(13)-dien-12-oic acid (5) Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}}$ cm⁻¹ 3500–2700, 1715, 1640 (C=CCO₂H), 1750, 1260 (OAc), MS m/z (rel int) 292 167 [M]⁺ (4) (calc for C₁₇H₂₄O₄ 292 167), 232 [M – HOAc]⁺ (63), 217 [232 – Me]⁺ (100), 171 (24), 91 (37)

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+42 \ +43 \ +51 \ +88} \,(\mathrm{CHCl}_{3}, \, c \ 0 \,9)$$

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